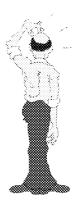


Introduction

What You Need to Know

- Regulations
- Principles of working with hazardous chemicals
- Standard lab practices
- Specific chemical information



Introduction

Regulations

- OSHA Regulations
 - Hazard Communication Standard
 - Lab Standard
 - PPE Standard
 - Air Contaminant Standards
- EPA, DOT, NRC
 - Resource Conservation & Recovery Act

Introduction

Dr. Karen Wetterhahn (1948-1997) Dartmouth College

- Dimethylmercury
- TLV = 0.01 mg/M³ (0.001 ppm)
- Potent neurotoxin
- Poisoning difficult to treat
- DMM experiments
- Latex gloves
- NO protection
- Poisoning 1st symptoms 3 mos., severe 5 mos.
- Died 10 months



General Rules

- Minimize all chemical exposures
- Know your chemicals & inventory
- Plan ahead
- Never underestimate the risks
- Prepare for emergencies







General Rules

- Apply good work practices
- Use PPE
- Apply good housekeeping practices
- Use ventilation & engineering controls to prevent/eliminate aerosol production



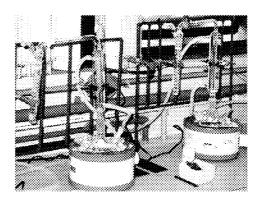
General Rules

- Minimize quantities used, ordered, stored
 - Smaller amounts = smaller risks
 - Reduces storage, waste disposal
- Use less toxic alternatives

General Rules

Know the Hazard

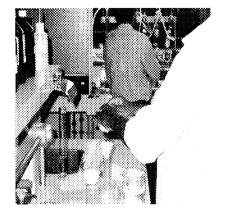
- Physical
 - * Explosive
 - * Flammable
 - Reactive
- Health
 - * Toxic
 - Corrosive



General Rules

Know Route of Exposure

- Contact
 - Skin
 - Eyes
 - Mouth
- Ingestion
- Inhalation
- Puncture/Cut



General Rules

Protect Yourself

- * Information
 - Labels
 - MSDS
 - Manufacturers fact sheets
 - Review work procedures



General Rules

MSDS (Material Safety Data Sheet)

- Chemical ID #
- Physical hazards
- Physical/chemical characteristics
- Reactivity
- Health hazards
- Controls PPE, hygiene, engineering
- Emergency response

General Rules

Practice good housekeeping

- Clean up after each experiment
- Don't store materials on work surfaces
- Keep aisles clear
- Keep chemicals in storage cabinets
- Purge work areas of unnecessary or unused equipment, supplies, chemicals

General

- 1. Access to the lab is restricted
- 2. Signs identifying hazards and responsible personnel posted on all access doors
- 3. PPE (lab coat/apron, eye wear, gloves) are required for ALL personnel and visitors in areas where chemicals are used and stored.
- 4. Eating, drinking, applying cosmetics, and food are not permitted in lab areas.

General

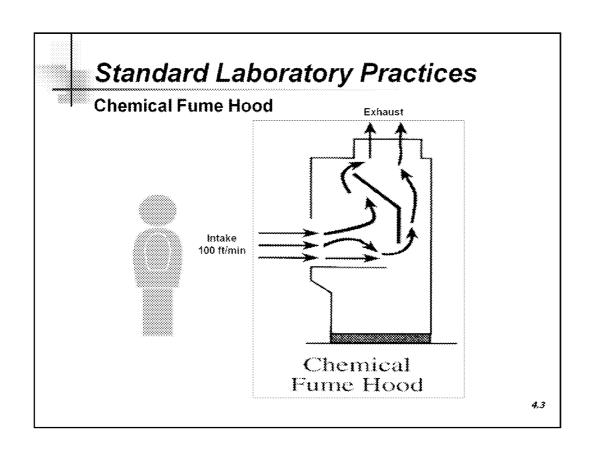
- 5. Mouth pipetting is prohibited
- 6. All chemicals are stored according to "compatibilities"
- 7. Chemical inventories are maintained and accessible
- 8. All chemical containers are labeled and dated after opening

General

- 9. Compressed gas cylinders are secured at all times
- 10. Chemical fume hoods, splash shields and/or respirators are used to prevent inhalation exposure
- 11. Outdated and obsolete chemicals are disposed of annually
- 12. Wastes are properly packaged and labeled for disposal

General

- 13. Spills are promptly cleaned up and area decontaminated
- 14. Work surfaces are cleaned regularly and kept uncluttered
- 15. Emergency equipment is checked regularly for operation and kept free from obstructions
- 16. All chemical laboratories have a lab specific chemical safety manual



Chemical Fume Hoods

Proper usage:

- Ensure hood certification annually
- Ensure hood working before use
- Work at least 6 inches inside
- Minimize storage



Chemical Fume Hoods

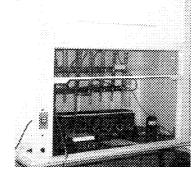
Proper usage:

- Place all materials into hood before beginning work
- Limit arm movements; move straight in and straight out
- Keep sash at certification point or lower
- Ensure slot at back of hood is not obstructed
- Elevate equipment above hood floor

Chemical Fume Hoods

Proper usage:

- Don't block air foil
- Run cables under air foil
- Don't put large equipment in hood
- Do not turn off hood when chemicals are inside



Chemical Fume Hood

Safe Practices

- Do Not use fume hoods to store chemicals
- Check with RSO before using radionuclides
- Limit activity in area when working at hood
- Always lower glass shield to certification mark or lower
- Perform work slowly move straight in and out

Hazardous Chemicals

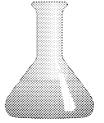
Flammables

- Fire, explosion most dangerous & expensive of accidents
- Reduce risk
 - Minimize quantities
 - Ventilation, fume hoods
 - No Ignition sources
 - Store in flammable cabinets, chemical safe refrigerators

Hazardous Chemicals

Corrosives

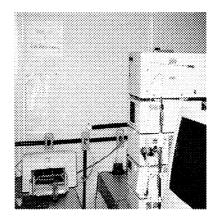
- PPE for skin, eyes
- Emergency eye wash/shower
- Incompatibilities
- Storage
- Containment of spills



Hazardous Chemicals

Toxic Chemicals

- Know your chemicals
- Ventilation, fume hoods
- **■** Contamination control
- PPE (specific)
- Minimize exposures



Hazardous Chemicals

Highly Toxic Chemicals

- Restricted access
- Experienced personnel
- Practice w/ non-hazardous chemical
- Contain
 - Ventilation
 - . PPE
 - Work practices
 - Proper disposal

- Reduce amount of chemical
- Segregate chemicals
- Label containers w/:
 - Receipt date; opening date; user name
- Know hazards of chemicals
- Know your inventory
- Discard chemicals on schedule

- Avoid storing chemicals on top of cabinets
- Keep exits, passageways, floors free of chemical containers
- Avoid storing chemicals in hoods, or on bench tops



- Store odorous chemicals in ventilated cabinets
- Store highly toxic chemicals in secondary containers
- Secure highly toxic chemicals



- Do not expose stored chemicals to:
 - Heat
 - Direct sunlight
- Do not generally store alphabetically
- Watch for & separate incompatibles



- Separate into compatible groups
 - Then store alphabetically
- Store in appropriate cabinets
 - Flammables only in approved cabinets
 - Flammable in laboratory-safe refrigerators
 - Acids, bases, oxidizers in separate cabinets

Flammables

- Class 1A: extremely flammable
 - f.p. below 23°C/b.p. below 38°C
 - diethyl ether, pentane, hydrogen
- Class 1B: ignites at normal temperature
 - f.p. below 23°C/b.p. at or above 38°C
 - Acetone, ethanol, toluene
- Class 1C: ignites when moderately heated
 - f.p. at or above 23°C and below 38°C
 - Styrene, xylene

Flammables

- Many serious accidents w/ storage in domestic refrigerators (not lab safe)
- All refrigerators, freezers for the storage must be lab safe
- NFPA 45 Standards for Lab Refrigerators

Peroxide Formers

- Among most hazardous chemicals
- **■** Form spontaneously
- Shock & heat sensitive
- **■** Explosive hazard!!

4,3

Peroxide Formers

- Date container
 - Upon receipt
 - At opening
- Discard all unopened
 - After 18 Months
- Never distill or concentrate
- Store in dark, cool place

Peroxide Formers (most dangerous)

- Discard 3 months after opening
 - Isopropyl ether
 - Divinylacetylene, butadiene
 - Vinylidene chloride, chloroprene, tetrafluoroethylene
 - Potassium metal
 - Sodium amide, potassium amide

Peroxide Formers (most used)

- Discard 12 months after opening
 - Diethyl ether, tetrahydrofuran, dioxane, other ethers
 - Acetaldehyde, acrolein, acetal, other aldehydes & acetals
 - 2-Propanol, 2-butanol, 2-pentanol, other secondary alcohols
 - Methyl isobutyl ketone

Peroxide Formers (frequently used)

- Discard 12 months after opening
 - Cumene, cyclohexene, dicyclopentadiene, alkenes
 - Acrylic acid, methyl methacrylate, acrylonitrile, other acrylics
 - Benzyl alcohol, 2-phenylethanol, tetralin, other benzylics
 - Decalin, methylcyclopentane, hydrocarbons w/ tertiary H's

Incompatibles

- React violently with each other to produce:
 - **■** Explosion
 - Heat
 - Highly toxic or flammable products or both
- Separate Incompatibles

Incompatibles (oxidizers)

- * React with:
 - organic chemicals, reducing agents, metals, hydrides, phosphorus, sulfur, carbon, ammonia
- Store solids together; liquids together

Incompatibles (oxidizers)

■ Solids:

 Perchlorates, peroxides, dichromates, nitrates, picrates, permanganates

■ Liquids:

 Nitric acid, perchloric acid, sulfuric acid, hydrogen peroxide, bromine

Incompatibles (acids)

- React with:
 - Bases, metals & their salts, hydrides, sulfides, azides, cyanides, nitrates, organic peroxides, oxidizing agents, water
- Store separately from other chemicals
- Separate organic acids & inorganic acids

- 1) Separate
 - Flammables, oxidizers, acids, bases
- 2) Separate
 - Organic & inorganic families
- 3) Separate
 - Families into related compatible groups
- 4) Then store alphabetically

Standard Laboratory Practices Chemical Waste

Safe Procedures:

- Do Not pour chemicals down the drain!
- Place waste in proper containers
- Dispose of chemical waste according to facility and local guidelines

Standard Laboratory Practices Chemical Waste

Creative Waste Handling

- Use alternate (less hazardous) chemicals
- Minimize quantities used
- Recycle chemicals
- Reduce the hazard before disposal
 - **■** Distillation
 - **■** Neutralization
 - **■** Evaporation

Personal Exposure

- **■** Don't panic
- Send for help
 - If you feel you cannot handle
 - If you need a respirator
- Assist any injured or contaminated personnel
 - Injured persons are priority
 - Obtain medical attention
- Notify supervisor



Eyes

- Immediately flush w/ water for 15 min
- Assist or get assistance during flushing
- Hold or assist in holding eye lids open
- Move eyeball around during flushing
- Get medical attention



Skin

- Flush immediately w/ flowing water no less than 15 min
- Remove jewelry in exposed area
- Wash w/ warm water & soap if no burn visible
- Check MSDS/other information for delayed effects
- Get medical attention



Clothing

- Use safety shower
- Quickly remove contaminated clothing, shoes, jewelry
 - Time is critical to prevent serious skin burns
 - Modesty must not delay removal of clothing; cut clothing off if necessary
 - Don't waste time wiping chemical from clothing or trying to neutralize it



Clothing (cont.)

- Be careful not to spread chemical from clothing to skin or eyes
- Flush contaminated areas w/ water for 15 min
- Get medical attention

Environmental Exposure

- Notify personnel
- Evacuate if necessary or unsure
- Control entry until spill cleaned up
- Call emergency response if necessary
- Call security to keep area clear & direct emergency response





Environmental Exposure

Spills

Use socks, pillows or booms for containment:

- Universal sorbent
 - Shredded polypropylene fill
- Use for solvents, oils, cautics, acids*
- Do not use w/:
 - Fuming sulfuric acid, fuming nitric acid
 - Strong oxidizers



Environmental Exposure

Solvents

- Use solvent absorbents
 - Solusorb
 - Vermiculite, clay, sand, kitty litter, oil Dri, granular activated carbon
- Use on common organics
- Do not use Solusorb for:
 - Strong oxidizers, including peroxides
 - Highly reactive organics



Solvents

- Remove all ignition sources
- Provide maximum ventilation.
- Wear appropriate PPE



Environmental Exposure

Acids

- Use acid neutralizers
 - Neutrasorb
 - Sodium carbonate or bicarbonate
- Use for common acids
 - Hydrochloric acid, perchloric acid, sulfuric acid, nitric acid, phosphoric acid, acetic acid

Environmental Exposure

Acids

- Cautions for Neutrasorb
- Do not use w/
 - Hydrofluoric acid, peroxy (per) organic acids, highly reactive chemicals
- Wear appropriate PPE.



Environmental Exposure

Caustics

- **■** Use caustic neutralizers
 - Neutracit-2
 - Sodium bisulfate, citric acid
- Wear appropriate PPE!
- Provide maximum ventilation

Environmental Exposure

Mercury

- Wear protective gloves
 - Latex, PVC, nitrile
- Use Hg Absorb sponges & powder
- Use flask w/ aspirator for small spills
- **■** Use Hg vacuum for larger spills
- Dispose of Hg waste properly
 - Keep in tightly closed containers

Environmental Exposure

Strong Oxidizers

- Use inert absorbents
 - Sand, clay (cat litter), vermiculite
- Caution:
 - Absorbed chemical still hazardous
- Wear appropriate PPE
- Dispose of as hazardous material